Three New Caged Prenylxanthones from Garcinia bracteata

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Three new caged prenylxanthones (xanthone $= 9H$ -xanthen-9-one), named neobractatin (1), 3-Omethylneobractatin (2), and 3-O-methylbractatin (3), along with eight known compounds, were isolated from the twig of *Garcinia bracteata*. The structures of the new compounds were elucidated on the basis of 1D- and 2D-NMR experiments, including HMBC, HSQC, ¹H,¹H-COSY, and ROESY, as well as HR-MS analysis.

Introduction. – The plants of the genus *Garcinia* (Guttiferae) have been extensively investigated from the phytochemical and biological points of view. Xanthones [1] [2], benzophenones [3], depsidones [4], flavonoids [5], biflavonoids [6], and triterpenes [7] have been isolated from African and southeast Asian species. Garcinia bracteata C. Y. Wu ex Y. H. Li is distributed in the south of Yunnan and Guangxi Province of P. R. China [8]. Previous phytochemical investigations on G. bracteata resulted in the isolation of caged prenylxanthones and benzophenones [9] [10]. Caged xanthones, having a rearranged skeleton, were mainly isolated from *Garcinia* species, such as G. bracteata, G. cantleyana [11], G. gaudichaudii [12], G. hanburyi [13], G. morella [14], and G. scortechinii [15]. As a part of our search for secondary metabolites from tropical plants, a careful investigation of the twigs of G. bracteata led to the isolation of three new caged prenylxanthones, neobractatin (1), 3-O-methylneobractatin (2), and 3-Omethylbractatin (3), together with five known caged prenylxanthones, bractatin (4) [9], isobractatin (5) [9], 1-O-methylisobractatin (6) [9], neoisobractatin A (7) [10], neoisobractatin B (8) [10] (*Fig. 1*), and three known 'classical' xanthones, gerontoxanthone I $[16]$, morusignin I $[17]$, and macluraxanthone $[18]$. The details of the isolation and structure elucidation of the new compounds, $1-3$, are reported in this article.

Results and Discussion. – Compound 1 was obtained as a yellow amorphous powder and showed an $[M + Na]$ ⁺ peak at *m/z* 487.2085 (calc. 487.2096) in the HR-ESI-MS, corresponding to a molecular formula of $C_{28}H_{32}O_6$. The IR spectrum exhibited a broad band at 3425 cm⁻¹ due to an OH group, and absorbtions at 1751 and 1637 cm⁻¹ due an unconjugated C $=$ O group and a chelated *ortho*-OH C $=$ O group, respectively. The ¹H-NMR spectrum (*Table 1*) revealed the presence of a chelated OH group (δ (H) 12.67 (s)), an olefinic H-atom of an α , β -unsaturated C=O unit (δ (H) 7.15 (d, J = 7.0 Hz)), three coupled H-atoms $(\delta(H) 6.30 (dd, J=10.0, 18.0), 4.81 (d, J=18.0),$ and 4.69 (d, $J = 10.0$), which were assigned to three olefinic H-atoms of a 1,1-

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Fig. 1. Structures of compounds 1 – 8

Table 1. ¹H-NMR Data of Compounds $1-3¹$). δ in ppm, J in Hz.

	1^{a} ^b)	2^c ^d)	3^c ^d)
$HO-C(1)$	12.67(s)	12.77(s)	13.27(s)
CH(2)	5.97 (s)	6.08(s)	6.11 (s)
$MeO-C(3)$		3.79(s)	3.76(s)
$H - C(7)$	3.77 $(dd, J=5.0, 7.0)$	3.72 $(dd, J=5.0, 7.0)$	3.48 $(dd, J=5.0, 7.0)$
$H-C(8)$	7.15 $(d, J = 7.0)$	7.16 $(d, J = 7.0)$	7.47 $(d, J = 7.0)$
$H - C(12)$	6.30 (dd, $J = 10.0, 18.0$)	6.32 (dd, $J = 10.0, 18.0$)	6.14 (dd, $J = 10.0, 18.0$)
CH ₂ (13)	4.81 $(d, J=18.0)$,	4.86 $(d, J = 18.0)$,	4.75 $(d, J=10.0)$,
	4.69 $(d, J = 10.0)$	4.75 $(d, J = 10.0)$	4.69 $(d, J = 18.0)$
Me(14)	$1.53 (s)^e$	$1.60 (s)^e$	$1.69 (s)^e$
Me(15)	$1.51 (s)^e$	$1.57 (s)^e$	$1.62 (s)^e$
$H_a - C(16)$	2.30 $(d, J = 14.0)$	2.51 $(d, J=13.0)$	2.49 $(d, J = 9.0)$
$H_b - C(16)$	1.84 $(dd, J=10.0, 14.0)$	1.78 $(dd, J=10.0, 13.0)$	
$H_a - C(17)$	2.20 $(dd, J=5.0, 10.0)$	2.16 (dd, $J = 5.0, 10.0$)	2.32 (dd, $J = 5.0, 13.0$)
$H_b - C(17)$			1.31 (dd, $J = 9.0, 13.0$)
Me(19)	1.21 (s)	1.33(s)	1.22(s)
Me(20)	1.25 (s)	1.35(s)	$1.65 (s)^e$
CH ₂ (21)	2.27 (dd, $J = 8.0, 14.0$),	2.48 $(dd, J=8.0, 14.0)$,	$2.64 - 2.62$ (<i>m</i>)
	1.84 $(dd, J=8.0, 14.0)$	2.06 (dd, $J = 8.0, 14.0$)	
$H - C(22)$	4.90 $(t, J = 8.0)$	5.01 $(t, J=8.0)$	4.37 $(t, J=8.0)$
Me(24)	1.63(s)	1.72(s)	1.39(s)
Me(25)	$1.52 (s)^e$	$1.58 (s)^e$	1.08(s)

^a) Recorded in (D_6) DMSO. ^b) Recorded at 400 MHz. ^c) Recorded in CDCl₃. ^d) Recorded at 500 MHz. e) Signals may be exchanged.

1) Arbitrary atom numbering. For systematic names, see Exper. Part.

dimethylprop-2-en-1-yl, and an olefinic H-atom of 3-methylbut-2-en-1-yl $(\delta(H)$ 4.90 $(t,$ $J = 8.0$)). The ¹³C-NMR spectrum (*Table 2*) exhibited 28 C-atom signals which were sorted by a DEPT experiment as those of six Me, three $CH₂$, and six CH groups, and of thirteen quaternary C-atoms (including two C=O C-atom signals at δ (C) 199.9 and 177.9, resp.). In the ¹H-NMR spectrum, characteristic signals at δ (H) 3.77 (dd, J = 5.0, 7.0, H-C(7))¹), 2.20 (dd, J=5.0, 10.0, H-C(17)), and 1.84 (dd, J=10.0, 14.0, $H_b - C(16)$) were discernible. Together with the presence of three O-bearing quaternary C-atom signals at $\delta(C)$ 83.6 (C(18)), 83.6 (C(10a)), and 78.9 (C(6)), and the evidence of correlations between CH₂(16) and C(5), C(7), C(8a), C(10a), and $C(17)$, correlations between H-C(17) with C(6), C(7), C(16), and C(18) in the HMBC spectrum $(Fig. 2)$ led to the assumption that 1 was a caged prenylxanthone, resembling 1-O-methylneobractatin [9], isolated from the same species, except for an OH group instead of a MeO group at $C(1)$. The C-atom signals at $\delta(C)$ 162.1 (C(1)) and 177.9 $(C(9))$ of compound 1 appeared at more down-field than those of 1-Omethylneobractatin (160.5 and 175.0, resp.) [9], due to the $HO-C(1)$ chelated with $C(9)=O$ in 1. Moreover, the HMBCs (*Fig. 2*) of HO $-C(1)$ ($\delta(H)$ 12.67 (s)) with C(1) $(\delta(C)$ 162.1), C(2) (97.1), and C(9a) ($\delta(C)$ 100.5) also ascertained the presence of an OH group at $C(1)$. ROESY Correlations (*Fig. 3*) were detected between $H-C(7)$, $H-C(17)$, $Me(19)$, $H-C(8)$, $CH₂(21)$, and $H-C(22)$, indicating that $H-C(7)$ and $Me(19)$ have the same orientation as the isoprenyl chain at $C(6)$. Therefore, compound 1 was elucidated as neobractatin.

	1^{a} ^b)	$2^c)^d$)	$3^c)^d$)		1^{a} ^b)	$2^c)^d$	$3^c)^d$)
C(1)	162.1	163.8	164.0	CH ₂ (13)	106.2	106.0	106.5
CH(2)	97.1	93.9	97.1	Me(14)	28.3	29.3	30.8
C(3)	167.0	168.1	168.1	Me(15)	27.4	28.9	27.9
C(4)	112.6	114.9	113.5	$CH2(16)$ or $CH(16)$	32.0	32.7	49.4
C(4a)	159.1	158.4	159.1	$CH2(17)$ or CH(17)	41.4	42.4	26.5
C(5)	199.9	199.6	84.9	C(18)	83.6	83.6	83.1
C(6)	78.9	79.3	204.1	Me(19)	26.5	26.7	29.2
CH(7)	44.5	44.8	47.4	Me(20)	29.2	29.6	31.2
CH(8)	135.5	134.0	134.0	CH ₂ (21)	30.1	30.2	28.8
C(8a)	133.2	134.8	133.3	CH(22)	117.6	117.4	118.0
C(9)	177.9	178.7	180.0	C(23)	135.4	136.1	135.1
C(9a)	100.5	102.0	101.5	Me(24)	17.9	18.1	16.9
C(10a)	83.6	83.6	91.4	Me(25)	25.7	25.8	25.5
C(11)	40.1	41.2	41.4	$MeO-C(3)$		55.4	55.3
CH(12)	151.0	151.4	150.7				

Table 2. ¹³C-NMR Data of Compounds $1-3^1$). δ in ppm.

Compound 2, a yellow amorphous powder, was shown to have the molecular formula $C_{29}H_{34}O_6$ on the basis of its HR-ESI-MS (m/z 501.2254 ($[M + Na]$ ⁺; calc. 501.2253)). The ¹H- and ¹³C-NMR (*Tables 1* and 2), HSQC, and HMBC spectra indicated that the structure of 2 was similar to that of 1, except for the presence of a

Fig. 2. Key HMBCs (H \rightarrow C) of compounds 1-3

Fig. 3. Key ROESY correlations $(H \leftrightarrow H)$ of compound 1

MeO group $(\delta(H)$ 3.79 (s); $\delta(C)$ 55.4) in 2 instead of the phenolic OH group in compound 1. The key HMBC correlation (*Fig. 2*) of MeO (δ (H) 3.79) with C(3) (δ (C) 168.1) indicated that the MeO group is located at $C(3)$. Thus, the structure of 2 was deduced as 3-O-methylneobractatin.

Compound 3 was isolated as a yellow amorphous powder with the same molecular formula $C_{29}H_{34}O_6$ as 2, deduced from the HR-ESI-MS (m/z 501.2264 ([$M + Na$]⁺; calc. 501.2253)). The NMR spectra (*Tables 1* and 2) of 3 suggested that its structure was very similar to that of 2. The main difference between the structures was the different position of the unconjugated C=O. Correlations from $H_b-C(17)$ ($\delta(H)$ 1.31) to the C=O (δ (C) 204.1), and from H_b-C(16) (δ (H) 1.78) to the C=O (δ (C) 199.6) were

observed in the HMBC spectrum of 3 and 2, respectively $(Fig. 2)$, which indicated that oxo group in 3 is located at $C(6)$, *i.e.*, $C(6)=O$. Hence, compound 3 is an isomer of 2, and its structure was established as 3-O-methylbractatin.

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Experimental Part

General. Column chromatography (CC): silica gel (SiO₂; 200 – 300 mesh; *Qingdao Marine Chemical* Inc., P. R. China), $SiO₂$ H (10 – 40 µm; *Qingdao Marine Chemical Inc.*), *Lichroprep RP-18* gel (40 – 63 mm; Merck, D-Darmstadt), MCI gel (75 – 150 mm; Mitsubishi Chemical Corporation, Japan), and Sephadex LH-20 (Pharmacia). Fractions were monitored by TLC, and spots were visualized by heating SiO₂ plates sprayed with 10% H₂SO₄/EtOH. Optical rotations: *Jasco DIP-370* digital polarimeter. UV Spectra: Shimadzu-UV-2401A spectrophotometer in MeOH solns.; λ_{max} (log ε) in nm. IR Spectra: Bio-Rad-FTS-135 spectrometer in KBr pellets; $\tilde{\nu}$ in cm⁻¹. 1D- and 2D-NMR Spectra: *Bruker-AM-400* and - $DRX-500$ spectrometers; δ in ppm rel. to Me₄Si as internal standard, J in Hz. MS: VG-Auto-Spec-3000 spectrometer with glycerol as matrix for FAB; $API-OSTAR-Pulsar-1$ spectrometer for HR-ESI; in m/z (rel. %).

Plant Material. The twigs of G. bracteata were collected from Xishuangbanna, Yunnan Province, P. R. China, in August 2008, and authenticated by Prof. Guo-Da Tao, Xishuangbanna Tropical Botanical Garden. A voucher specimen (No. 20080801) was deposited with the ethnobotany research group of Xishuangbanna Tropical Botanical Garden, Chinese Academy of Sciences.

Extraction and Isolation. The air-dried twigs of G. bracteata (6.5 kg) were powdered and then extracted with 95% EtOH (3×201) at r.t. The combined solns, were concentrated to dryness under vacuum. The crude extract was suspended in H_2O and successively extracted with petroleum ether (PE) and AcOEt. The combined PE extracts were evaporated to give a deep-brown gum (154 g), which was separated on a SiO₂ column, eluted with a gradient of PE/AcOEt $95:5-40:60$ to afford six fractions, Frs. $A - F$. Fr. C (8.6 g) was further chromatographed (SiO₂; PE/AcOEt 4:1) to provide 1 (35 mg), 4 (65 mg), and five subfractions, *Frs. C1 – C5. Fr. C2* was purified by reversed-phase (RP) CC (*RP-18*; MeOH/H₂O 7:3-9:1) to give 5 (28 mg) and 2 (22 mg). Fr. C3 was subjected to RP-18 CC (MeOH/H₂O $6:4-8:2$) to afford 6 (32 mg) and 3 (26 mg). Fr. D (3.5 g) was fractionated by CC (SiO₂; PE/AcOEt 3:1), followed by Sephadex LH-20 (MeOH), to provide 7 and 8 (mixture, 42 mg). Fr. E (6.5 g) was divided into three subfractions by CC (SiO₂; CHCl₃/MeOH 30:1): *Frs. E1 – E3. Fr. E1* was further purified by CC ($RP-18$; MeOH/H₂O 7:3) to yield gerontoxanthone I (19 mg) and morusignin I (26 mg). Fr. E3 was purified by CC (Sephadex LH-20; MeOH) to provide macluraxanthone (11 mg).

Neobractatin $(=(+)-(3R*,3aS*,10aR*,11aR*)-9-(1,1-Dimethylprop-2-en-1-yl)-2,3,3a,11a-tetrahy$ dro-6,8-dihydroxy-2,2-dimethyl-11a-(3-methylbut-2-en-1-yl)-3,10a-methano-10aH-furo[3,2-b]xanthene-5,11-dione; **1**). Yellow amorphous power. $[\alpha]_0^{27} = +9.4$ ($c = 0.24$, MeOH). UV: 208 (4.59), 346 (4.19). IR: 3425, 2920, 1751, 1637, 1592. ¹H- and ¹³C-NMR: *Tables 1* and 2, resp. FAB-MS (pos.): 465 ($[M+H]^+$). HR-ESI-MS: 487.2085 ([$M + Na$]⁺, C₂₈H₃₂NaO₆⁺; calc. 487.2096).

3-O-Methylneobractatin $(=(+)-(3R^*,3aS^*,10aR^*,11aR^*)-9-(1,1-Dimethylprop-2-en-1-yl)-$ 2,3,3a,11a-tetrahydro-6-hydroxy-8-methoxy-2,2-dimethyl-11a-(3-methylbut-2-en-1-yl)-3,10a-methano- $10aH$ -furo[3,2-b]xanthene-5,11-dione; 2). Yellow amorphous power. $\left[\alpha\right]_0^{26} = +5.4$ (c = 0.17, MeOH/ CHCl₃). UV: 237 (4.19), 348 (4.28). IR: 3432, 2923, 1751, 1634, 1588. ¹H- and ¹³C-NMR: *Tables 1* and 2, resp. ESI-MS: 501 ($[M + Na]^+$). HR-ESI-MS: 501.2254 ($[M + Na]^+, C_{29}H_{34}NaO_6^+$; calc. 501.2253).

3-O-Methylbractatin $(=(-)(1S*,3aR*,5R*)-11-(1,1-dimethylprop-2-en-1-yl)-3,3a,4,5-tetrahydro-8-1-d)$ hydroxy-10-methoxy-3,3-dimethyl-1-(3-methylbut-2-en-1-yl)-1,5-methano-1H,7H-furo[3,4-d]xanthene-7,13-dione; 3). Yellow amorphous power. $\lbrack a \rbrack_0^2 = -1.3$ (c = 0.44, MeOH). UV: 217 (4.51), 354 (4.15). IR: 3443, 2930, 1736, 1637, 1581. ¹H- and ¹³C-NMR: *Tables 1* and 2, resp. ESI-MS: 501 ($[M + Na]^+$). HR-ESI-MS: 501.2264 ($[M + Na]^+$, C₂₉H₃₄NaO₆⁺; calc. 501.2253).

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